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A HYPOGLYCEMIC AGENT

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Specification

1. Title of Invention

Hypoglycaemic agent

2. Patent Claim

Hypoglycemic agent which has a compound represented by the following formula as the active component.

$$CON < \frac{R_1}{R_2}$$

[In the formula, R₁ denotes hydrogen atom or lower alkyl group, R₂ denotes a linear, branched or cyclic alkyl group, a pyridyl group which may have a substituent on the nucleus or a pyridylmethyl group, and n denotes 1-3].

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3. Detailed Description of the Invention

This invention is the invention of a hypoglycemic agent which has a compound represented by the following formula (I) as the active component

[In the formula, R₁ denotes hydrogen atom or lower alkyl group, R₂ denotes a linear, branched or cyclic alkyl group, a pyridyl group which may have a substituent on the nucleus or a pyridylmethyl group, and n denotes 1-3].

Known compounds are included in the aforesaid compound represented by the formula (I), but in the previous literature in which they are mentioned, there is no mention at all of a hypoglycemic effect or a pharmacological action suggesting this.

The compounds of this invention represented by the aforesaid formula (I) may be obtained readily by usual methods of reacting an amine compound with a methoxybenzoyl chloride compounds in the presence of a base such as triethylamine, as illustrated in the following reference example.

Reference Example

4-methoxybenzoyl chloride 17 g was added gradually under ice cooling and stirring to a mixed solution of 3-aminopyridine 9.4 g, triethylamine 15ml and acetone 200 ml. After stirring for 30 minutes at the same temperature then for 60 minutes at room temperature, the reaction solution was poured into 11 of water, and the crystals which precipitated were collected by filtration and washed with water, then re-crystallised from methanol, to obtain 175 g of colourless acicular crystals of 4-methoxy-N-3-pyridylbenzamide (compound 1), melting point 168-170°C.

Elemental analysis as molecular formula $C_{13}H_{12}N_2O_2$ $C \qquad H \qquad N$ theoretical value (%) 68.41 5.30 12.27
experimental value (%) 68.33 5.27 12.24

The compounds of Table 1 were obtained in the same way.

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Table 1

No	-(OMe)n	R_1	R_2	Molecular formula	Melting point (°C)	-	d Eleme Calc 6) Found	(%) C	H	nes N N
2	2-0Ne	н		0 1 2H 1 2N 2 O 2	112~114	7 6	6 8.4 1 6 8.4 9	5.3 0 5.2 4	1 2.2 7].
3	•	,	-Q _{He}	O14H14N2O2	.80~82	8 3	6 9.4 0 6 9.3 2	5.83 5.80	1 1.5 6 1 1.5 9	
. 4 .	,	•	Ö.,	O16H16N2O2	85~87	9 1	7 0.3 9 7 0.2 4	6.29 6.23	1 0.9 3 1 0.9 9	
5	3-OMe	•		O13H11N2O2	121~122	8 5	6 8.4 1 6 8.4 B	5.3 0 5.3 6	1227	
6	,	•	Q	,	155~156	8 9	6 8.4 1 6 8.4 3	5.3 0 5.3 1	1227 1230	
7	. ,	,	√) _N	O14H14N2O2	99~101	8 8	6 9.4 0 6 9.4 7	5. B 3 5. 7 9	1 1.5 6 1 1.6 0	7
8	4-OMe	•		OtaHiaNaOa	191~182	7 9	6841 6835	5.3 0 5.2 6	1227	
9		•	- cu, C	O14H14N2O2	150~153	6 5	6 9.4 0 6 9.3 6	5.8 8 5.7 9	1 1.5 6 1 1.5 2	
10	•	,	- CM2 - CM3	,	71~73	6.8	6 9.4 0 6 9.4 7	5.8 3 5.7 8	1 1.5 6 1 1.5 8] ·
1.1	•	•	√µ_n₄	,	61~64	7 7	6 9.4 0 6 9.4 5	5.83 5.88	1 1.5 6 1 1.6 3	
1 2	•	•	٥٠	O15H14N2O2	136~137	8 2	7 0.2 9 7 0.3 7	6.29 6.34	1093 1089	

			· · · · · · · · · · · · · · · · · · ·		<u>.</u>				
1 3	2,3-(OMe) ₂	н		O14H14N2O3	117~118	5 8	6 5.1 0 6 5.1 4	5.46	1 0.8 5
1 4	•	•	€ _N L _{He}	O15H16N2O3	110~111	6 2	6616	5.9 2 5.9 5	10.29
1 5	,		Ŏ	. O1 6H1 8N2O2	111~112	6 7	6 7.1 1 6 7.1 4	6.34	9.78 9.75
16	2,4-(OMe)2	•	-cn2-CN	O15H16N2O3	98~99	5 1	66.16	5.92 5.87	10.29
17		•	-Q-ne		140~141	69	6 6.1 6 6 6.2 1	5.9 2 5.9 6	10.29
1 8	•.	•	₩.	O14H14N2O3	93~94	6 3	6 7.1 1 6 7.1 5	6.34	9.78
1 9	2,6 - (OMe)2		- Q	O15H16N2O3	155~156	6 7	66.16	5.92 5.97	1029
2 ()	•	,	Ö,ne	O16H18N2O3	206~209	6 3	67.11	6.34	9.78
2 1	3,4 -(OMe)g	,		O14H14N2O3	84~86	7 9	6 5.1 0 6 5.1 6	5.4 6 5.4 1	1 0.8 5
2 2	,	,		,	49~51	8 8	6 5.1 0 6 5.0 8	5.4 6 5.4 3	1 0.8 5
2 3	,	,	-cu ₂	O18H16N2O2	122~123	6 3	6 6.1 6	5.9 2 5.9 7	1 0.8 8 1 0.2 9 1 0.2 4
2 4	,	,	- 647	,	128~129	7.4	6 6.1 6 6 6.1 9	5.9 2	1 0.2 9
2 5	,	,	Q _{Ne}	,	131~132	7 5	6 6.1 6 6 6.2 0	5.88 5.92 5.96	1 0.3 3

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2 6	3,4-(OMe)2	н	He He	O16H18N2O2	69~71	63	6 7.1 1 6.3 4 9.7 8 6 7.1 5 6.3 7 9.7 7
27	,	•	i-Pr	O12H17NO3	144~145	8 5	6 4.5 5 7.68 6.27 6 4.5 9 7.6 1 6.2 3
2 8	,	,	#-Bu	O13H19NO3	83~84	88.	6 5.8 0 8.0 7 5.9 0 6 5.7 8 8.0 3 5.8 4
2 9	•	•	s - B u	,	1 2 7~1 2 8	8 3	65.80 8.07 5.90 65.84 8.04 5.93
3 0		•	i -Bu		124~125	80	65.80 8.07 5.90 65.85 8.11 5.95
3 1	•	•	-(1)	OssHaiNOs -	181~182	9 1	6841 8.04 5.32 6836 8.07 5.36
3 2.	3,5-(OMe) ₂	,	- N He He	O18H16N2O3	96~97	8 5	66.16 5.92 10.29 66.12 5.98 10.32
33.	,	,	, in	O18H18N2O3	119~120	8 7	6 7.1 1 6.34 9.78
3 4	3,4,5-(OMe) ₃	•	\Diamond	O18H16N2O4	154~156	6 5	6 2 4 9 5.5 9 9.7 2
3 5	•	•	Q	•	157~158	7 7	6249 559 9.72
3 6	÷ ,		- 01,	O16H16N2Q4	115~116	5 8	6356 600 9.27
3 7	•	•	-01,	,	145~146	6 9	6352 6.04 9.25 6356 6.00 9.27 6351 6.07 9.22
3 8	•	,	ne ne	,	127~128	6 4	6 3:5 6 6.00 9.2 7 6 3:5 9 6.03 9.2 9

3 9	3,4,5-(OMe);	н	Q _n	017H26N2O4	145~146.	71	6 4.5 4 6 4.5 8	6.37	8.8 6 8.9 0
40	•	•	n-Pe	O19H19NO4	114~115	73	6 1.6 4 6 L6 0	7.5 6 7.5 9	5.5 3 5.5 7
4 1	,		i-Pr	•	154~155	77	6 L6 4 6 L6 6	7.5 6 7.5 4	5.5 3 5.5 8
4 2	. ,	,	#-Bu	014H21NO4	133~134	80	6290	7.9 2 7.8 6	5. 2 4 5. 2 7
4 3	•		e-Bu	•	162~163	7 5	6 2 9 0 6 2 9 5	7.9 2 7.9 4	5. 2 4 5. 2 0
4.4	•	•	t - Bu		133~134	7 9	6290 6291	7.9 2 7.8 8	5.24 5.29
4 5	•	•	i-Bu	•	122~123	8 1	6290 6296	7.9 2 7.87	5,24 5,28
4 6	•		~(H)	018H23NO4	182~183	8 8	6 5.5 1 6 5.5 4	7.9 0 7.9 3	4.7 8
47		i-Pr	i-Pr	O18H26NO4	127~128	7 2	6 5.0 6 6 5.1 1	8.5 3 8.5 9	474

The compounds of this invention obtained in this way have excellent hypoglycemic action, and are effective at 100 mg/kg in man, and their effect is maintained by administration of 0.1-100 mg once a day for 24 hours or more.

For administration, a preparation is used which has been formed into the desired form by a customary means normally used in drug formulation.

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Example 1

5-week-old mice (male, body weight 25-30g) with 5 animals in a group were fasted for 16 hours, and then alloxan at 75 mg/kg was administered intravenously. After 48 hours, a solution or suspension of a compound of this invention (200 mg/kg) was administered orally, and after 150 minutes, blood was taken from the heart and the glucose level was measured using glucose oxidase. The measurement results are exemplified in Table 2.

Table 2

Administered	Blood glucose value (mg/dl)
compound	$mean \pm S.D.$
None (control)	473 ± 28
1	3 2 6 ± 4 2 ••
3	3 7 8 ± 3 1 ••
4	3 6 4 ± 1 9 •••
6	3 7 8 ± 5 2 •
7	4 1 2 ± 3 3 •
1 2	3 8 3 ± 2 8 ••
1 7	3 4 5 ± 4 1 ***
2 2	3 7 8 ± 3 7 ••
2 5	355±46 **
2 6	3 3 6 ± 3 2 •••
2 7	407±30•
2 8	402 ± 24 ••
2 9	4 2 1 ± 2 7 • ;·
3 2	4 1 6 ± 2 3 •
3 3	4 0 2 ± 3 4 •
3 6	4 1 6 ± 2 1 ••
3 8	3 0 7 ± 4 3 •••
3 9	4 1 2 ± 3 1 •
4.1	4 2 1 ± 2 8 •
4 6	383±41.00

In the Table, the compound number corresponds to the compound number of the reference examples.

Example 2

4-methoxy-N-3-pyridylbenzamide (compound 1)	100 parts
calcium hydrogen phosphate	58.5 parts
crystalline cellulose	50 parts
corn starch	40 parts
calcium stearate	1.5 parts

These components were mixed well and pressed into 250 mg tablets (content of active component 100 mg/tablet) by usual methods, for use as a hypoglycemic agent.

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